## SESEY Summer Experience in Science and Engineering for Youth

#### **Big Picture:**

Anammox provides a cheap, less energy intensive Nitrogen removal process for wastewater treatment. Why remove nitrogen? Eutrophication!

### What We Did

- I. Made media
- 2. Made standard curves for NH4+ and NO2- and determined concentrations for standard (y=mx+b) with colorimetry.
- 3. Dilute influent to lower concentration to linear range.
- 4. Measured flow rate (Q) mL/min to determine mass removed.

All in order to treat wastewater and save energy

## Ammonium (NH<sub>4</sub>+)

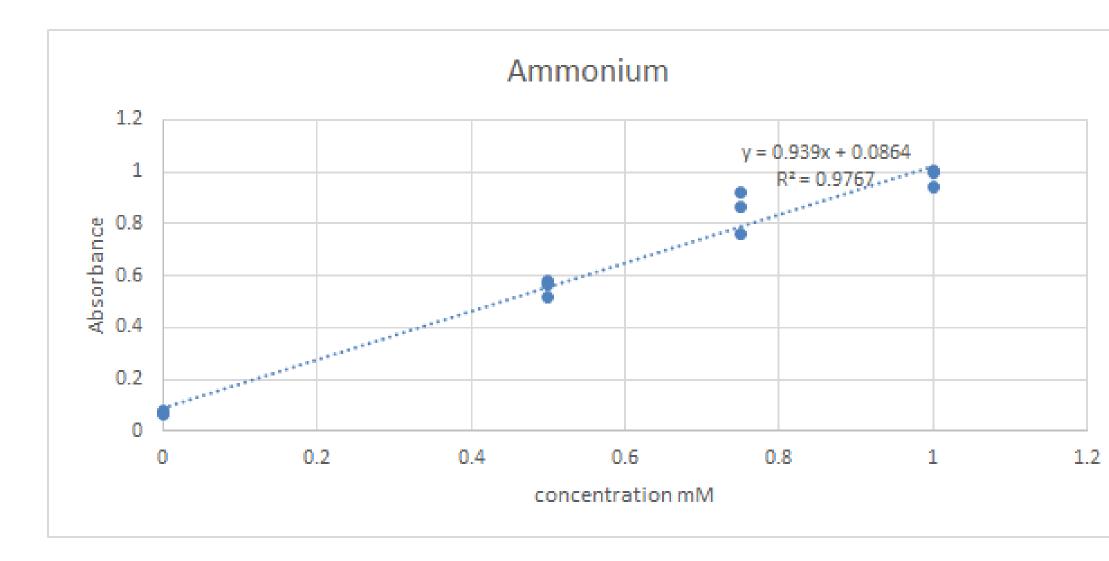
#### **Procedure:**

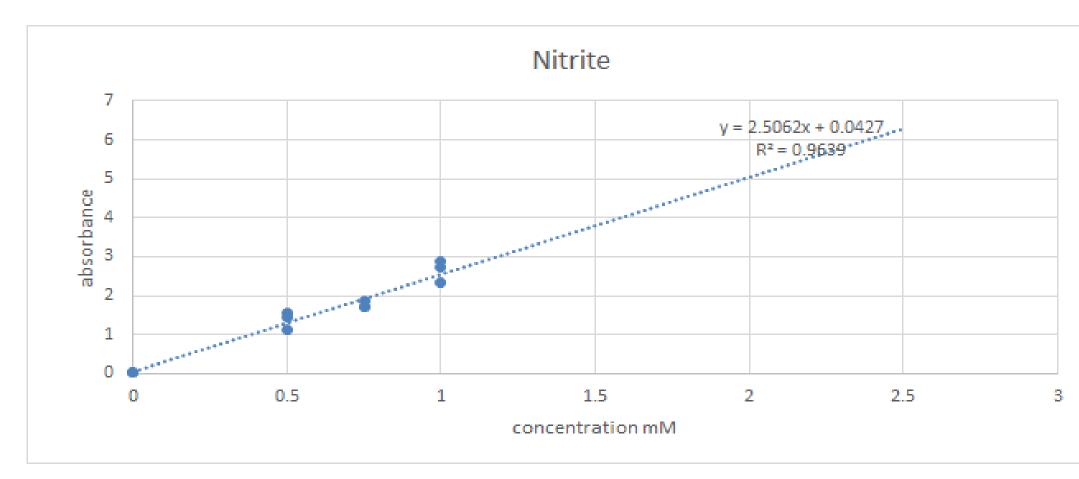
Add 25 uL of sample Add 175 uL of Citrate reagent Let sit for 1 minute Add 50 uL of 2-Phenylphenol-Nitroprusside reagent Add 25 uL of Buffered Hypochlorite reagent Complete these steps for each of the absorbance levels Put into 37 degree Celsius incubator for 15 minutes Measure absorbance at 660 nm Put waste into labeled amber bottle Standard curve of data is graphed below

# Nitrite $(NO_2^-)$

#### **Procedure:**

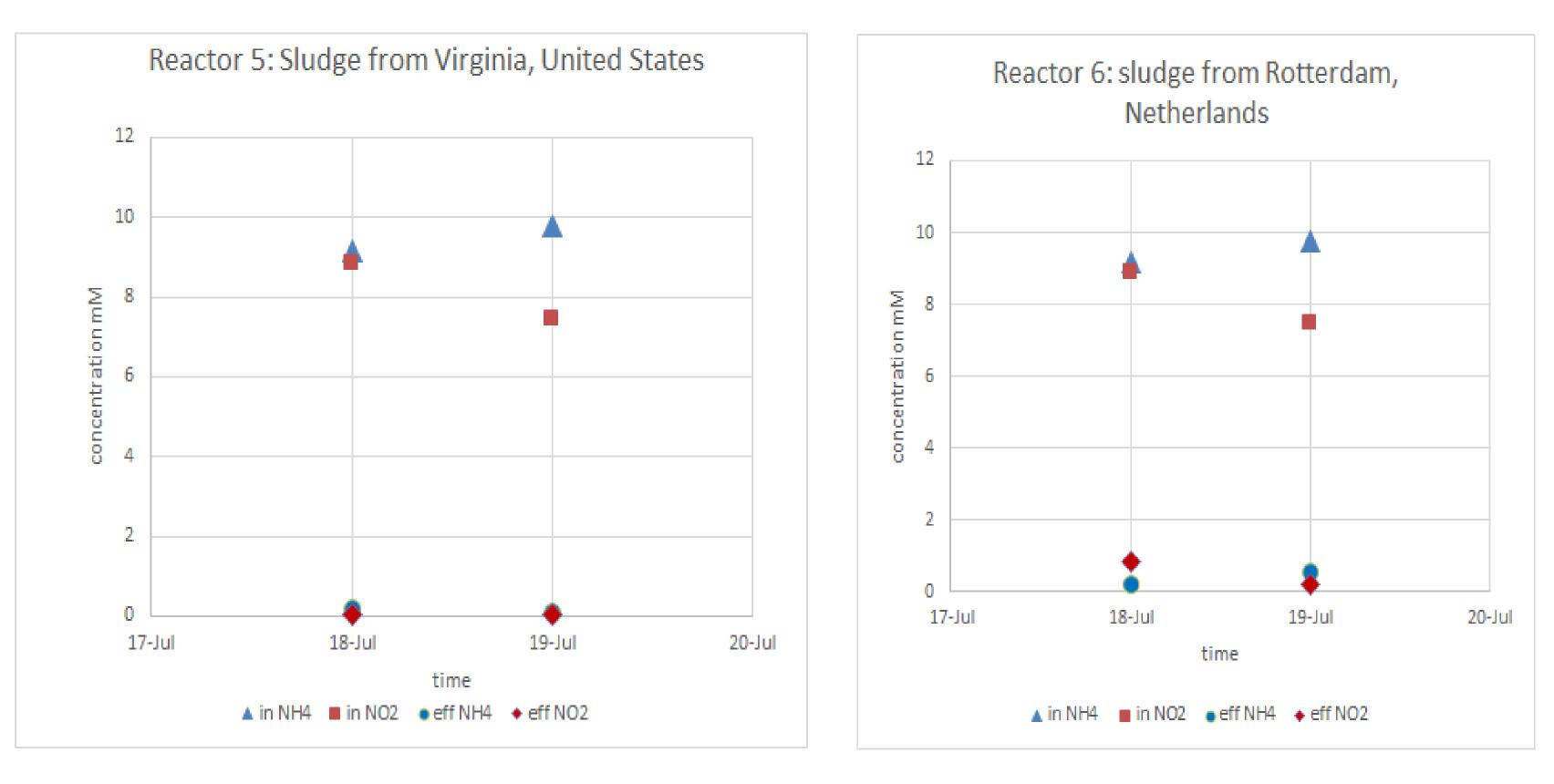
Add 200 uL of the sulfanilamide solution Add 20 uL of sample Add 20 uL of 0.1% NED reagent Mix by pipetting each well up and down Wait 10 minutes Measure absorbance at 540 nm Standard curve of data is graphed below

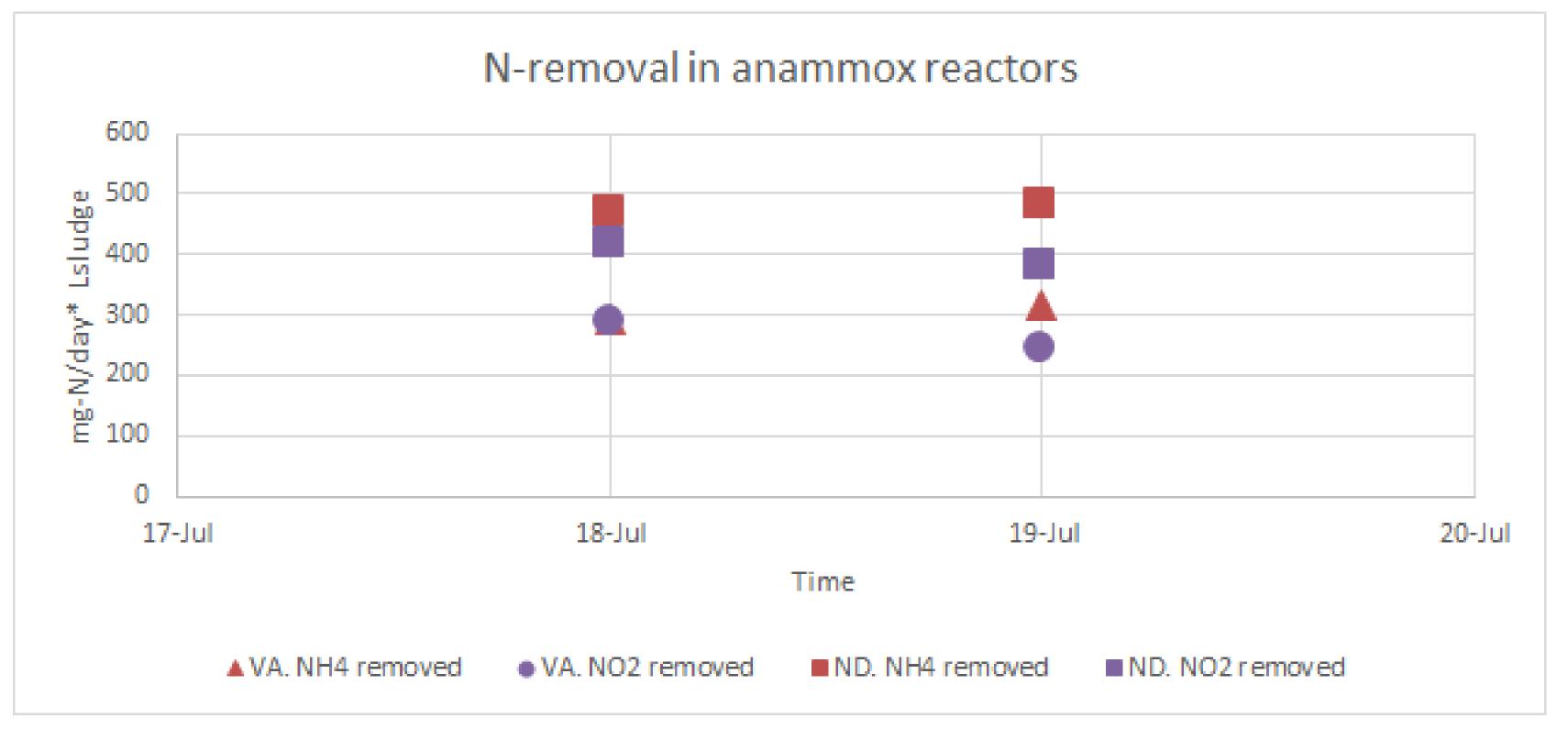




#### Why the ANAMMOXity? Julia Pingel, Deven Leon Patino, Rich Hilliard, Dr. Tyler Radniecki at Oregon State University, Environmental Engineering Vocabulary to know

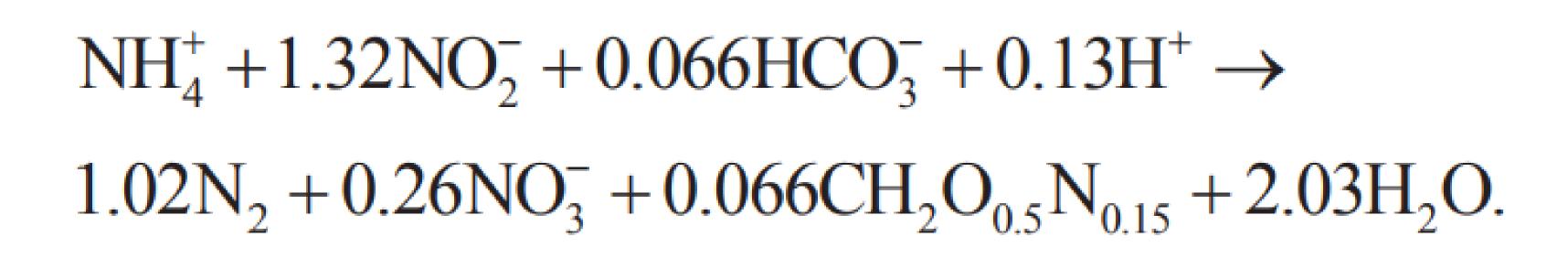
**Results:** 





#### **Conclusions:**

Nitrogen removal was determined to be **571 mg-N/day\*L sludge** in the Virginia reactor and 875 mg-N/day\*L sludge in the Netherlands reactor. These results do not account for nitrate in the mass balance as the assay uses a highly corrosive acid. However, these results are consistent with the historical observations for these reactors. From the equation below (van de Graaf et al. 1997) one can see that this represents the production of **22 mmol Nitrogen gas/day\*L sludge** in the Virginia reactor and **34 mmol Nitrogen gas/day\*L sludge** in the Netherlands reactor.



Anammox biomass equation van de Graaf et al. 1997



- Anammox = ANaerobic AMMonium OXidation
- *Eutrophication* = excessive richness of nutrients

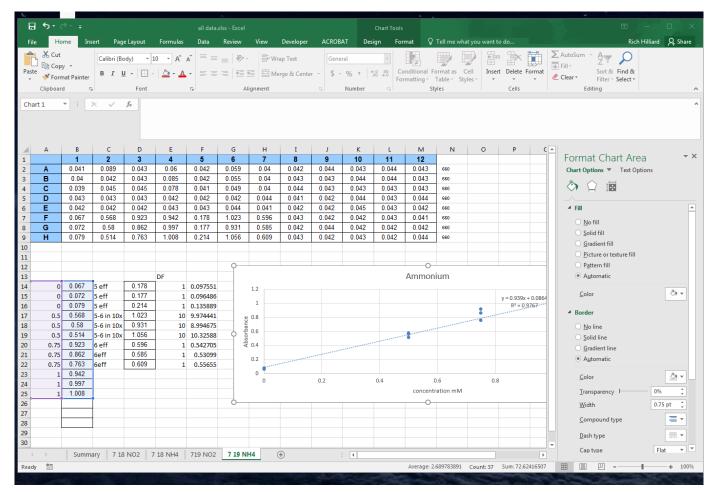
#### Materials:

Ammonium Colorimetric Assay-2-Phenylphenol Tetrahydrate Sodium Nitroprusside Trisodium Citrate Sodium Phosphate Sodium Hypochlorite Solution Nitrite Colorimetric Assay-Sulfanilamide N-(1- Naphthyl) Ethylenediamine Dihydrochloride Hydrochloric acid NaNO 2 DIH2O

#### Learning Outcomes:

To the right is some sludge where the anammox lives. Try and find the nitrogen gas bubbles!

Below is our Excel spreadsheet where we analyzed our data.



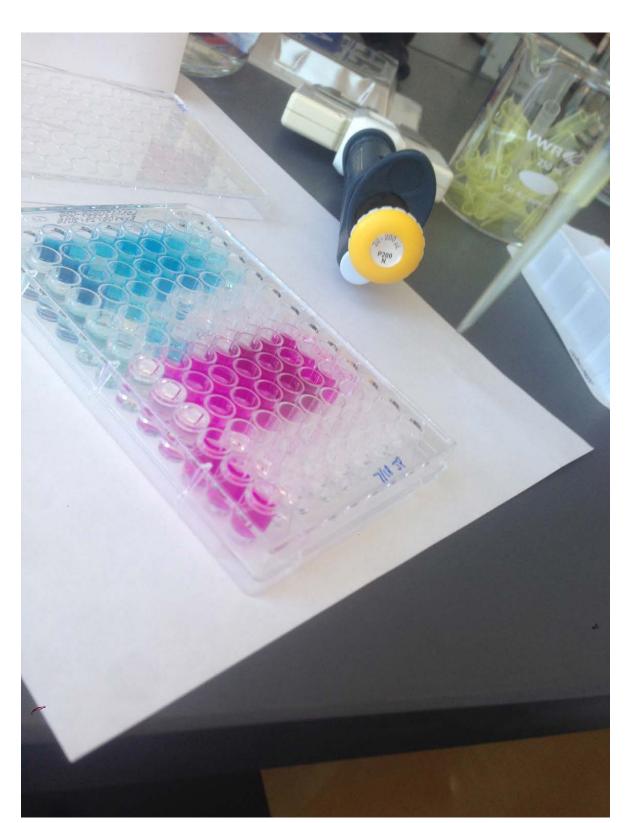


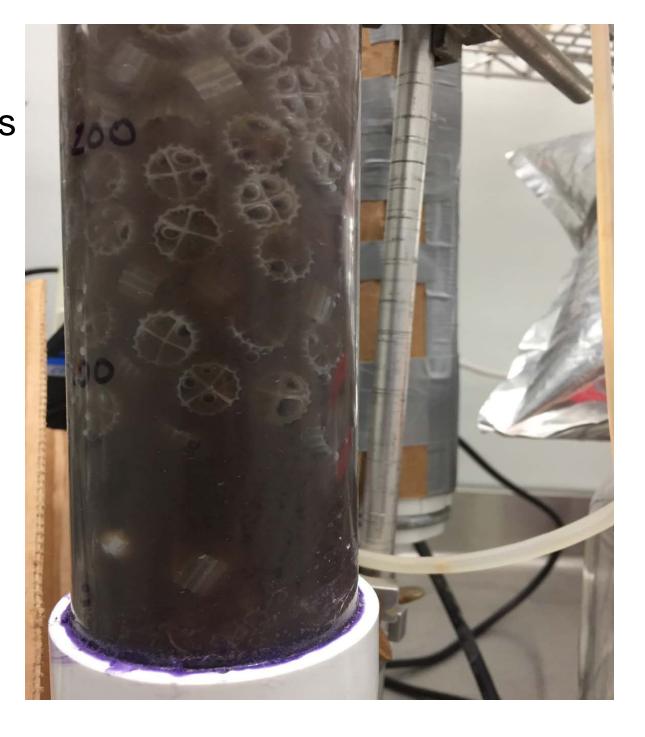
# **Acknowledgements:**

Rochefort



<u>Media = A liquid designed to support the growth of microorganisms or cells</u>





Left to Right: Rich Hilliard, Julia Pingel, Deven Leon Patino posing comfortably in the walk in refrigerator while modeling proper PPE (personal protective equipment!).

We would like to thank Oregon State University and Dr. Skip